

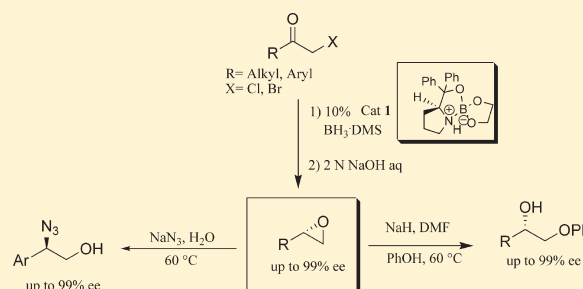
Chiral Epoxides via Borane Reduction of 2-Haloketones Catalyzed by Spiroborate Ester: Application to the Synthesis of Optically Pure 1,2-Hydroxy Ethers and 1,2-Azido Alcohols

Kun Huang, Haiyang Wang, Viatcheslav Stepanenko, Melvin De Jesús, Carilyn Torruellas, Wildeliz Correa, and Margarita Ortiz-Marciales*

Department of Chemistry, University of Puerto Rico-Humacao, CUH Station, Humacao, Puerto Rico 00791

Supporting Information

ABSTRACT: An enantioselective borane-mediated reduction of a variety of 2-haloketones with 10% spiroaminoborate ester **1** as catalyst is described. By a simple basic workup of 2-halohydrins, optically active epoxides are obtained in high yield and with excellent enantiopurity (up to 99% ee). Ring-opening of oxiranes with phenoxides or sodium azide is investigated under different reaction conditions affording nonracemic 1,2-hydroxy ethers and 1,2-azido alcohols with excellent enantioselectivity (99% ee) and in good to high chemical yield.



Nonracemic epoxides are recognized not only for their biological importance,¹ but also as key intermediates in diverse chemical transformations for the synthesis of chiral auxiliaries, natural products, and chiral drugs.^{2,3} Moreover, chiral epoxides have been recently used as catalysts for asymmetric induction in the autocatalyzed organozinc addition to an aldehyde.⁴ Consequently, in the last 30 years, a variety of direct epoxidation and resolution methods for the synthesis of enantiopure oxiranes have been successfully developed.^{5–8} More convenient and efficient routes to prepare optically pure styrene oxides involve the initial enzymatic⁹ or chemical asymmetric reduction of α -halogenated acetophenones.^{10–12} Environmentally friendly boron reagents have been used for the reduction of a variety of α -haloketones and the halohydrins are easily converted to nonracemic epoxides without racemization.^{11,12} These methods can be considered, essentially, as “one pot” procedures valuable for substrates with sensitive groups due to the very mild reaction conditions. The most recognized chiral catalysts for the enantioselective borane-mediated reduction of unsymmetrical ketones with outstanding and predictable stereoselectivities are 1,3,2-oxazaborolidines (OAB) derived from nonracemic 1,2-amino alcohols.¹² Nevertheless, the in situ prepared *B*-H oxazaborolidines are known to form side products diminishing in some cases the effectiveness of the catalyst.¹³

A series of stable crystalline aminoborate esters, derived from a variety of nonracemic 1,2-amino alcohols, ethylene glycol, and inexpensive triisopropyl borate, were previously reported by our group.^{14,15} The most effective and convenient aminoborate complexes used as catalysts for the borane-mediated reduction of ketones^{14,15a–15g} and benzyl oximes^{15h} are shown in Figure 1.

Spiroaminoborate ester **1** derived from diphenylprolinol was highly effective for the reduction of a variety of aryl,^{15a–15d} heteroaryl,^{15e} α -alkoxy,^{15f} and aliphatic^{15g} ketones providing the desired secondary alcohols in excellent yield with excellent enantioselectivity, higher or similar to Corey’s OAB (CBS reagent). Presently, we are interested in the synthesis of nonracemic α -aryloxy and α -amino alcohols as important chiral building blocks for the preparation of neurobiological active molecules. We, herein, describe a simple and convenient procedure for the preparation of optically active epoxides in high yield with excellent enantiopurity and their application to the synthesis of optically pure β -hydroxy ethers and 1,2-azido alcohols.

To develop a suitable procedure for the synthesis of epoxides, the asymmetric reduction of α -chloroacetophenone was initially carried out with 10% catalyst **1** and 0.7 equiv of $\text{BH}_3 \cdot \text{DMS}$ in THF at room temperature, followed by treatment of the crude chlorohydrin with 2 N NaOH. The styrene oxide was afforded in good yield and without racemization. To optimize the reaction conditions, α -Cl and α -Br acetophenone were reduced under different catalytic loads and borane sources, and subsequently, the crude halohydrins were cyclized under basic conditions. Since $\text{BH}_3 \cdot \text{DMS}$ provides similar enantioselectivity as $\text{BH}_3 \cdot \text{THF}$, we consider its use more attractive for a convenient method due to its low price and stability. In addition, the reduction of α -Br acetophenone afforded the corresponding epoxide in higher enantiopurity than the chloro analogue, providing 99% ee, and good chemical yield with 10% of catalyst **1**, 0.7 equiv of $\text{BH}_3 \cdot \text{DMS}$ in THF at room temperature for 0.5 h.

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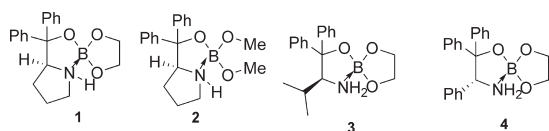


Figure 1. Representative aminoborate complexes.

Encouraged by the remarkable enantioselectivity obtained in the synthesis of the styrene oxide using spiroaminoborate **1**, we explored its scope in the reduction of a variety of aromatic and aliphatic α -halogenated ketones and their halohydrins conversion to enantioenriched oxiranes. As indicated in Table 1, α -haloacetophenones with Me, MeO, Cl, F, CN, and NO₂ ring-substituents (entries 1–10) were reduced to the *S*-2-bromo-1-phenylethanols, which were transformed under basic conditions to the *S*-epoxides in good to excellent yields and outstanding enantioselectivities (97–99% ee). The presence of different substituents on the aromatic ring has no significant effect on the ee values (entries 2–10). However, the cyclization of 2-bromo-4-methylphenyl-1-ethanol (entry 9) by treatment with 2 N NaOH afforded the corresponding epoxide in lower yield. However, the use of *t*-BuOK led us to the expected epoxide in good yield. In addition, the α -bromocyclohexyl ethanone and α -bromo adamantyl ethanone afforded the epoxides in high yield and excellent enantiomeric excess (entries 11 and 12).

The absolute configuration of the oxiranes was determined by comparing the optical rotation with the literature values.^{8,10,11a,15f,16a} Racemic β -hydroxy ethers were prepared as standards for HPLC analysis by the sodium borohydride reduction of β -phenoxyketones (**8**).

Regioselective ring-opening of oxiranes with nucleophiles is an important route to obtain a broad range of nonracemic 1,2-substituted alcohols as intermediates for the synthesis of key pharmaceutical compounds, as well as chiral catalysts.^{3,17–19} Optically active β -hydroxy ethers are valuable synthetic intermediates in a wide variety of potentially bioactive compounds.^{2g,17a} Although asymmetric syntheses of terminal epoxides in high optical purity are well established,^{5–8} efficient methods for the synthesis of nonracemic β -aryloxy alcohols via epoxide ring-opening remains limited.¹⁸ Thus, we were interested in studying the regioselective ring-opening of enantiopure epoxides with phenols or phenoxides. Initially, the ring-opening of racemic 2-phenyloxirane was studied by using sodium phenoxide in the presence of β -cyclodextrin in water at 60 °C.^{18c} However, the major product was 2-phenoxy-2-phenylethanol (**9a**), produced by ring-opening at the benzylic position contrary to the reported 2-aryloxy-1-phenylethanol (**10a**). The compound **9a** was identified by ¹H NMR analysis and compared with a standard sample of **10a** prepared according to previous reported procedure.^{15f} Further analyses of the reaction mixture by ¹H NMR showed **9a**, **10a**, and diol **11** in 100/8/88 ratio, and in 46%, 6%, and 38% isolated yield, respectively. For comparison purposes, other substrates were also reinvestigated. See the Supporting Information.

The low regioselectivity and poor yields in the previous methods led us to search an alternative route for the synthesis of the desired (*S*)-1,2-aryloxy ethanols **10**. After various optimization studies with different bases (K₂CO₃, NaOH, KOH, and NaH) and solvents (CH₃CN, THF, and DMF), in addition to various temperatures and reaction times, it was found that the treatment of epoxides **7** with phenol, previously treated with NaH, in anhydrous DMF at 60 °C for 36 h, provided the desired

Table 1. Asymmetric Reduction of α -Haloketones for the Synthesis of Epoxides^a

entry	epoxides	yield (%) ^b	ee (%) ^c	
1		7a	90	99
2		7b	90	99
3		7c	66	98
4		7d	89	99
5		7e	95	97
6		7f	78	95
7		7g	91	99
8		7h	76	99 ^d
9		7i	71 ^e	98 ^d
10		7j	72	98 ^f
11		7k	88	97 ^d
12		7l	93	>99 ^d

^a All the reactions were performed as reported in the experimental procedure. ^b Purified by flash chromatography or Kugelrohr distillation. ^c Determined by GC on a chiral column. ^d Determined by chiral HPLC on the basis of β -hydroxyl ether. ^e *t*-BuOK was used in the epoxidation reaction. ^f Determined by chiral HPLC.

(*S*)- β -hydroxy ethers **10** in good yield without racemization. As indicated in Table 2, representative aromatic and aliphatic (*S*)- β -hydroxy ethers were produced under the selected conditions with outstanding enantiopurity (97% to >99% ee) and in moderate to good isolated yields of pure products (entries 1–6).

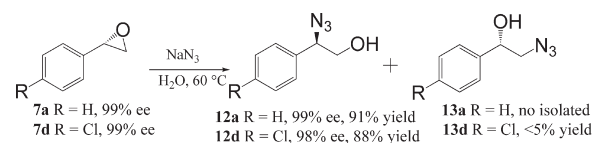
Optically active 2-azido-2-aryl ethanols are well-known precursors for the synthesis of a broad variety of enantiopure 1,2-amino alcohols, which are essential building blocks in the preparation of neurobiologically active compounds.^{3,17} Recently, Qu and co-workers^{20a} reported the preparation of racemic 1,2-diols and 1,2-azido alcohols by the ring-opening of epoxides in hot water. Due to the importance of nonracemic 1,2-amino alcohol, we were interested in studying the ring-opening reaction of optically pure epoxides with NaN₃. As indicated in Scheme 1, when styrene epoxide (**7a**) and sodium azide were heated in

Table 2. Preparation of β -Hydroxy Ethers 10 from Epoxides 7

entry	epoxides	yield (%) ^b	ee (%) ^c	
1		7a	90	99
2		7b	90	99
3		7c	66	98
4		7d	89	99
5		7e	95	97
6		7f	78	95
7		7g	91	99
8		7h	76	99 ^d
9		7i	71 ^e	98 ^d
10		7j	72	98 ^f
11		7k	88	97 ^d
12		7l	93	>99 ^d

^aThe products were obtained by preparative TLC. ^bDetermined by HPLC (Chiralcel OD-H). ^cReaction time 36 h.

water at 60 °C for 3.5 h, 2-azido-2-phenyl ethanol **12a** was formed as the main product with only a trace amount of regioisomer **13a**, as determined by GC-MS and ¹H NMR analyses of the crude product. The ratio of **12a** and **13a** by GC-MS analysis was 95.4:4.6. After purification by preparative TLC, it was afforded (*S*)-**12a** in 91% yield and 99% ee, based on HPLC analysis on a chiral column. Since highly enantiopure (*S*)-1-phenyl-2-azido ethanol was obtained, and more significantly, with complete inversion of configuration, the reaction was carried out with the *p*-chlorostyrene oxide **7d** to demonstrate the generality of this valuable ring-opening reaction. According to the analysis by GC-MS, the ratio of 2-azidoarylethanol **12d** to **13d** was 93:7. Remarkably, **12d** was isolated in excellent yield (88%) and its optical purity was 98% ee. The stereochemistry of these compounds was confirmed by full spectroscopic analysis and optical rotation. Interestingly, this transformation only slightly decreased the enantioselectivities of compounds **12a** (99% ee) and **12d** (98% ee), obtaining the 2-azido-1-arylethanol

Scheme 1. Ring-Opening of Epoxides with NaN₃ in Hot Water

with inversion of configuration at the stereogenic center. Noteworthy, this highly regioselective and enantioselective conversion of styrene oxides to 1,2-azido alcohols was conducted under mild and environmentally favorable conditions with only hot water, providing a rapid access to important non-natural amino alcohols, as well as to biological building blocks that can be used for example in the click chemistry for the formation of chiral triazoles or amides.²¹

In summary, the enantioselective reduction of 2-haloketones with 10% of the spiroaminoborate ester **1** offers a convenient and facile route for the synthesis of a variety of highly enantiopure epoxides. The ring-opening of racemic styrene oxides by aryloxy nucleophiles in hot water in the presence of β -cyclodextrin takes place by a preferential nucleophilic attack to the benzylic position providing 2-aryloxy-2-aryl ethanol as the mayor product. However, the treatment of chiral epoxides with PhONa in anhydrous DMF at 60 °C for 24–36 h gives mainly 1-aryl and 1-alkyl β -aryloxy ethanol **10** in outstanding ee and good chemical yield. In addition, remarkable regioselective nucleophilic attack of NaN₃ to (*S*)-epoxides in hot water takes place affording optically pure (*R*)-2-azido-2-aryl ethanol **13** with complete inversion of configuration in outstanding enantiopurity (up to 99% ee) and in high yield.

EXPERIMENTAL SECTION

General Procedure for Asymmetric Reduction and Epoxidation. To a solution of spiroaminoborate ester **1**¹⁵ (64 mg, 0.2 mmol) in dry THF (5 mL) was added BH₃·DMS (0.14 mL, 10 M, 1.4 mmol) at room temperature under N₂ atmosphere, and the solution was stirred for 30 min. A solution of the 2-haloketone (2 mmol) in dry THF (4 mL) was added dropwise for 1 h. After the addition was complete, the reaction mixture was stirred for 30 min, cooled at 0 °C, and quenched with MeOH (5 mL). The solvents were removed under reduced pressure, and the residue was treated with 2 N NaOH (5 mL) in THF (10 mL) for 2 h. After an aqueous workup the residue was purified by flash column chromatography on silica.

Typical Procedure for the Preparation of Enantiopure β -Hydroxy Ethers: (*S*)-(+)-1-(3-Methoxyphenyl)-2-phenoxyethanol (10j**).** To a dried, three-necked, 25 mL, round-bottomed flask charged with NaH (30.3 mg, 1.2 mmol, 1.2 equiv, 95%) in anhydrous DMF (2 mL) was added dropwise PhOH (94.1 mg, 1.0 mmol) in anhydrous DMF (2 mL) at ambient temperature under nitrogen. After the mixture was stirred for 10 min, a solution of epoxide **7j** (150.2 mg, 1.0 mmol) in anhydrous DMF (2 mL) was added dropwise. The mixture was stirred at 60 °C for 24 h and after aqueous workup the residue was purified by preparative TLC. The desired product was obtained in 82% yield (200 mg) as a colorless oil; 98% ee, determined by chiral HPLC analysis.

General Procedure for the Preparation of 2-Azido-2-arylethanol. To a stirred suspension of epoxide (1 mmol) in distilled water (6 mL) in a 25 mL flask with condenser was added sodium azide (2 mmol, 130 mg). The mixture was stirred for 3.5 h at 60 °C and then 20 mL of water was added. The product was isolated by preparative TLC.

■ ASSOCIATED CONTENT

Supporting Information. Characterization data of all compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

■ AUTHOR INFORMATION

Corresponding Author

*E-mail margarita.ortiz1@upr.edu

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